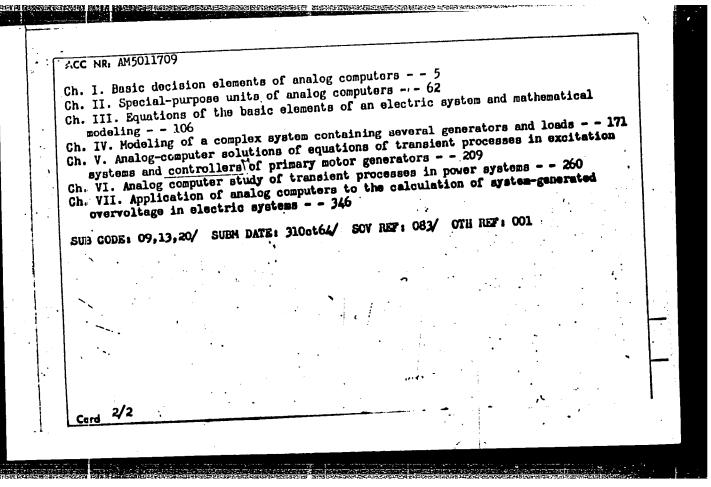
L 2968-66 ACCESSION NR: AP5026355

Moskovskiy rayon and the chief engineer in construction of the Bobrikovsk (now Nowomoskowsk) hydroelectric station. In connection with the reorganization of construction in 1931, Toleshev was transferred to Energostroy, first as chief engineer of the Moscow division and then as deputy chief of the design administration of Energostroy (now Teploelektroproyekt). In 1934. Teloshew took the post of assistant director of the Scientific Section, of the Power Engineering Institute imoni Krzhizhanovskiy of the Academy of Sciences USSR and worked as the immediate assistant to Academician G. M. Krzhizhanovskiy in directing the Institute until 1946. Starting in 1923, he did scientific research work first at the Moscow Institute of Mechanics im. Lomonosov and then at the Institute of National Economy in. Plakhanov. After the founding of the Moscow Power Engineering Institute in 1930. Toleshev transferred to that Institute and worked there until 1940. Here he was Lecturer of the Department of "Central Electric Stations" and a professor in the department. He received his professorship in 1933. He was Dean of the Sleatric Power Department of the Institute from 1932-1935. In 1940, Teleshev was made director of the Department of Electrical Engineering of the Moscow Institute of Fine Chemical Technology where he remained until 1955. In 1944 he took part in organizing the Power Engineer-

Ordahonikidae. From 1942 the Department of "Electritive printings of his texts ing." Teleshev has acted number of electrical statiment Consultation on the D has been an active member. Power Industry for more the form of the Society from 1944 been made a permanent member of Lenin. the Order of the Orig. art. has: 1 figure. ASSCCIATION: none.	there have been trical Engineer- mas for a great end in the Covern- y. I. Lenin. He Society of the f the Moscow board Society, he has uty in the Moscow d with the Order	2
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以对这种种的现在分词,也可以是一种的一种,但是是一种的一种的一种的一种的一种的一种的一种的一种的一种的一种的一种的一种的一种,但是一种的一种的一种的一种,可以一种 MONOGRAPH ACC NR. AM5011709 Gruzdev, Igor' Aleksandrovich; Kadomskaya, Kira Panteleymonovna; Fuchumov, Leonid Aleksandrovich; Luginskiy, Yakov Natanovich; Portnoy, Marlen Gealevich; Sokolov, Nikolay Ivanovich Using analog computers in power systems; methods for analyzing transient processes (Primeneniye analogovykh vychislitel'nykh mashin v energeticheskikh sistemakh; metody issledovaniy perekhodnykh protesessov) Moscow, Izd-vo "Energiya", 1964. 407 p. illus., biblio. 5,000 copies printed. TOPIC TAGS: analog computer, electromagnetism, electric engineering, electric power engineering, mathematic model, computer circuit, computer application PURPOSE AND COVERAGE: This book is concerned with the application of analog computers to the study of electromechanical and electromagnetic transient processes in power systems. It presents methods for mathematical modeling, circuits for special-purpose devices used in general-purpose computer studies, and examples of completed investigations. The book is intended for engineers at scientific research and planning institutes, workers at power systems, and students taking advanced courses in electric power and electromechanics. TABLE OF CONTENTS [abridged]; Foreword - - 3 UDG: 681.142.33/.34:620.9 Cord 1/2



SOURCE COME: 02/01/3/66/000/006/0128/0128 ACC NR: AP7002308

AUTHOR: Borchaninov, G. S.; Sokolov, N. I.; Vasil'yev, A. A.; Tarasov, Y. I.; Grudinskiy, P. G.; Ul'yanov, S. A.; Kuvshinskiy, N. M.; Fedoseyev, A. M.

ORG: none

TITLE: L. N. Baptidanov (Deceased) SOUF.CE: IVUZ. Energetika, no. 6, 1966, 128

TOPIC TAGS: electric engineering personnel, academic personnel ABSTRACT: L. N. Baptidanov died January 13, 1966. His working life was primarily dodicated to training of electrical engineering specialists. Soon after graduating from the Electrical Industrial Faculty of the Moscow Institute of the National Economy, Baptidanov began teaching at the Moscow Power Technical School. In 1934, Eaptidanov bogan teaching at the All Union Correspondence Industrial Institute, then in 1946 he shifted to the All Union Industrial Academy of Machine Building, where he worked in the chair of electrical power stations. He was responsible for the creation of a model electrical station in the electrical stations chair of the Moscow Power Institute. Baptidanov was also very active as an author, writing such works as "Industrial Enterprise Substations", "Electrical Equipment of Electrical Stations and Substations", etc. From 1943 to 1946, Baptidanov worked as the Scientific editor for Electrical engineering at the State Power Literature Publishing

House. [JPRS: 37,564]

SUB CODE: 09 / SUBM DATE: none

Card 1/1

SOURCE CODE: UR/0105/66/000/009/0019/0024 AP7006045 ACC NR: AUTHOR: Sokolov, N. I. (Doctor of technical sciences; Moscow); Kozlova, V. F. (Engineer; Moscow); Khvoshchinskaya, Z. G. (Engineer; Moscow) TITLE: Problems of stability of parallel operation of hydroelectric stations with capsule generators SOURCE: Elektrichestvo, no. 9, 1966, 19-24 TOPIC TAGS: hydroelectric power plant, electric generator, circuit breaker, ABSTRACT: Capsule generators, single horizontal shaft hydraulic turbines and generators placed in metal capsules, have poorer electromechanical characteristics than vertical hydraulic generators of the same unit power due to hydraulic and mechanical considerations. This article analyzes problems of static and dynamic stability of hydroelectric stations with capsule generators, problems of the selection of the most effective systems for excitation and automatic control of excitation. The installation of capsule generators considerably hinders provision of stability of parallel operation of stations and reduces the overall level of stability of combined power systems. All known methods of increasing stability are as yet little effective in this case. The most effective, though very expensive, method for increasing stability is the usage of controlled synchronous compensators. The best compromise for solving the entire problem is the usage of high speed circuit breakers and relays with overall operating times of under 0.06 seconds. Orig. art. has: 6 figures. [JPRS: UDC: 621.311.2.016.35 39,548] SUB CODE: 10, 09 Card 1/1

SOURCE CODE: UR/0104/66/000/008/0695/0096 ACC NR. A07007595 AUTHOR: Chuprakov, N. M.; Borovoy, A. A.; Postnikov, N. A.; Malychev, A. A.; Magidson, E. M.; Sin'chugov, F. I.; Zeylldzon, Ye. B.; Barchaninov, G. S.; Yermolenko, V. M.; Vasil'yev, A. A.; Sokolov, N. I.; Ul'yanov, A. S.; Fedeseyev, A. M.; Sarkisov, M. A.; Rokotyan, S. S.; Azar'yev, D. I.; Arson, G. S.; Dubinskiy, L. A.; Zhulin, I. V.; Kolpakova, A. I.; Antoshin, N. N. Kriaunchik, A. J.; Kachkin, M. D.; Preobrazhenskiy, N. Ye.; Reut, M. A.; Kneyfits, M. E.; Sharov, A. N.; Yakub, Yu. A.; Gorbunov, N. I.; Shurmukhin, V. A.; Beschinskly, A. A. ORG: none TITLE: Boris Sergeyovich Uspenskiy (on his 60th birthday) SCURCE: Elektricheskiye stantsii, no. 8, 1966, 95-96 TOPIC TACS: hydroelectric power plant, electric engineering personnel. SUB CODE: 10 ABSTRACT: 3. S. Uspenskiy was born in June 1906. He graduated from the State Electric Machine Building Institute in 1928 as an electric installation engineer. He worked in the State Electro-Technical Trust for four years, then in the Ali-Union ElectroTechnical Union, where he planned power construction units. Plans which he made up at that time for the electrical portion of electrical stations and sub-stations are still being used. He was invovied in planning and installation of the electrical portion of hydro-electric power stations and powerful pumping stations in the Mascow-Volga Canal. Euring the war, he was in charge in installation of the Krasnogroskaya Heat and Electric Power Station, the planning of the Urals Hydro-Electric Power Station and other projects. He Card 1/2 _____

44.0

SELIMOV, Midat Abdurakhmanovich. Prinimal uchastiye GORDON, Ya.Ya., prof.; SOKOLOV, N.I., red.; MATVEYEVA, M.M., tekhn.red.

[Ways of eliminating hydrophobia] Puti likvidatsii gidrofobii. Moskva, Medgiz, 1963. 293 p. (MIRA 17:2)



MELI'NIKOV, Nikolay Ivanovich; SOKOLOV, N.I., red.; PETROVA, N.K., tekhn. red.

[Pathogens of suppurative diseases and their association]Voz-buditeli gnoinykh zabolevanii i ikh assotsiatsii. Moskva, Medgiz, (MIRA 15:9)

1962. 263 p. (SUPPURATION) (BACTERIA, PATHOGENIC)

REZNIKOVA, Lyusi Solomonovna; EPSHTEYN-LITVAK, Rakhil' Veniaminovna; LEVI, Moisey Iosifovich; SOKOLOV, N.I., red.; LYUDKOVSKAYA, N.I., tekhn. red.

[Serological methodology of research in the diagnosis of communicable diseases] Serologicheskie metody issledovaniia pri diagnostike infektsionnykh boleznei. Moskva, Medgiz, 1962. (MIRA 16:3) 370 p. (SERUM DIAGNOSIS) (COMMUNICABLE DISEASES)

MATVEYEV, Konstantin Ivanovich; SOKOLOV, N.I., red.; SENCHILO, K.S., tekhn. red.

[Epidemiology and prevention of tetamus] Epidemiologiia i profilaktika stolbniaka. Moskva, Gos. izd-vo med. lit-ry Medgiz, 1960. 337 p. (MIRA 14:7)

"APPROVED FOR RELEASE: 08/25/2000 CIA-RDP86-00513R001652020004-1 A STREET OF THE PROPERTY OF TH

- PANINA, A. V. SOKOLOV, N. I.,
- 2. USSR (600)
- 4. Sand, Foundry
- 7. Reconditioning used molding sand by separation in a stream of air. Lit. proiz. no., 9, 152.

_1953. Unclassified. 9. Monthly List of Russian Accessions, Library of Congress, January

SOKOLOV, N.K.; KAPKO, P.S., kand.sel'skokhozyaystvennykh mauk; MALYUGINA, Ye.A., nauchnyy sotrudnik

Valuable mineral feed for swine. Svinovodstvo 13 no.11:26-28 N '59.

1. Glernyy zootekhnik Krasnodarskogo plodoovoshchnogo sovkhoza
No.2 (for Sokolov). 2. Krasnodarskiy nauchno-issledovatel'skiy
institut sel'skogo khozyaystva.

(Swine--Feeding and feeds) (Minerals in food)

(Sugar industry--By-products)

SOKOLOV, N.L.; SMUROV, A.M.

Power parameters for the extrusion of rod-type forgings with various shapes of flanges. Avt.prom. 29 no.12:31-34 D '63. (MIRA 17:4)

1. Nauchno-issledovatel'skiy tekhnologicheskiy institut avtomobil'noy promyshlennosti.

SOMOLOV, E. L.

USSF/Miscellaneous - Tools and dies

Card. 1/1

Pub. 12 - 7/12

Authors

Sokolov, N. L.

Title

The manufacture of the second Modification of swage die insets (chucks)

Periodical

8 Avt. trakt. prom. 4, 24-27, Apr 1954

fortract

Modifications of die insets (changes in inset sizes and configurations), for the purpose of economizing on expensive alloyed steel used in the manufacture of dies, are described. The changes were first introduced at the Low-Liter Capacity Auto Plant in Moscow and the gains derived from these modifications, are listed. Drawings.

Institution: The Low-Liter Capacity Auto Plant, Moscow

Submitted :

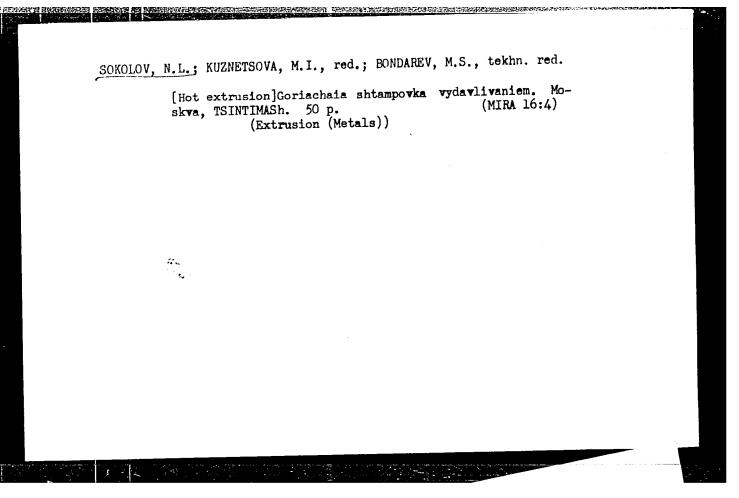
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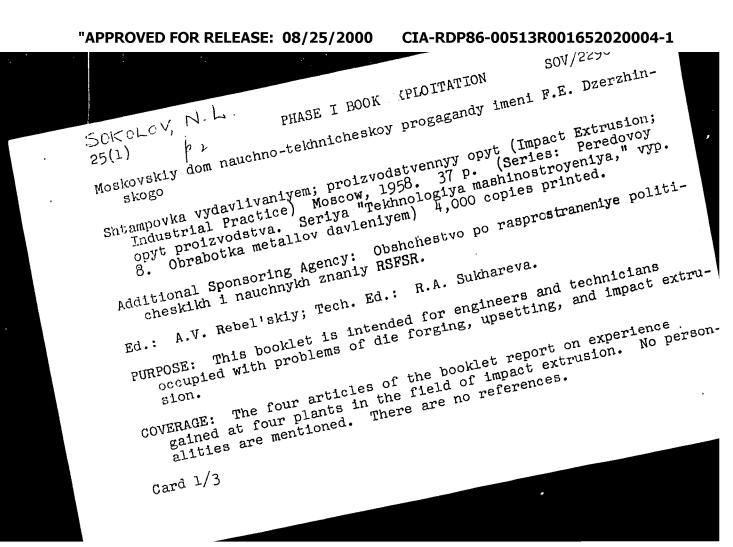
SOKOLOV, Nikolay Leonidovich; SAMOKHOTSKIY, A.I., inzh., ved. red.;

L'VOV, D.S., kand. tekhn. nauk, red.; SOROKINA, T.M.,
tekhn. red.

[Economy of die steel in drop forging with mechanical forging presses] Ekonomiia shtampovoi stali pri shtampovke na krivoshipnykh kovochno-shtampovochnykh pressakh. Moskva, Filial Vses. in-ta nauchn. i tekhn. informatsii, 1958. 11 p. (Peredovoi nauchno-tekhnicheskii i proizvodstvennyi opyt. Tema 5. No.M-58-202/11) (Forging) (MIRA 16:3)

(Dies (Metalworking))--Maintenance and repair)





Impact Extrusion; Industrial Practice

sov/2290

26

- Kozlov, I.N. Die Forging and Impact Extrusion From Ball-shaped Blanks TABLE OF CONTENTS: (Experience of the Pervyy gosudarstvennyy podshipnikovyy zavod [First
- The advantages of using ball-shaped blanks for impact extrusion (hot or cold) of ring-and cup-shaped steel parts are stressed, State Bearing Plant]) and arrangement of dies, materials used and technique are dis-
- Savin, G.P. Fabrication of Automobile Engine Valves by Direct Impact Extrusion With a Single Blow (Experience of the Gor'kly avtozavod
- The technique of the process and the dies and meterials used [Gor'kly Automobile Plant]) are described. The relationship between the speed of the ram and the speed of the flow of material is discussed.
- Sokolov, N.L. Making Steel Forgings by Hot Impact Extrusion (Experience of the Moskovskiy zavod malolitrazhnykh avtomobiley
 - The impact extrusion technique for making small, medium, and [Moscow Small Automobile Plant]) large forgings and the design of dies are described. Sugges-

Card 2/3

D FUR RELEASE: U8/25/2000 CIA-RDP80-00513KU01652020004-1

- Alervais, A.l., brim, T.D., Birteenolat, B.L., Gelevneve, L.A., Rievnet, Brom recembrish, Brance, Petr Bacimirovish, hutsovohip, F.L., Lavatship, T.M., Bekelov, B.L.
- Descripting a shrapper has (flashess Preso-Terming) Lucaev, Lashes, 1600 2500 1.00 copies wrinted.
- na.(withe pape): belowney, I.P., Candidate of Technical aciences; mayimates: Stellmakey, E.H. Engineer, and Eduardoy, M.J., Engineer; ad. (inside wish): decidayey, E.T., Engineer; Ed. of Publishing House: Chies, M.A.; Icon. Ed.: Speranskeys, C.W.; Managing Ed. for literature on the technology of machine building (Lenin rad Division of Eashgis): Managov, Me.P., Engineer.
- The cook is intended for engineering personel and it may be useful to students of vinces and technical schools.
- October: The cokpresents the processes of press forming without than in closed dies from steel and nonferrous alloys later called

Janc 1/2

Flashless Press-forming

892

flashless press-forming. The following suggestions for mastering this process are made: technical and economical indices, rules for designing parts to be made by this process, determining heating regimes preventing scale formation, methods of designing and cutting blanks, determination of capacity of forging equipment, design and calculation of dies, and reference tables. Typical production examples are included (with calculation and drawings for dies) and new data on flashless press forming techniques abroad are presented. There are 32 references of which 21 are Soviet and 11 are English.

Card 2/2

CIA-RDP86-00513R001652020004-1 "APPROVED FOR RELEASE: 08/25/2000

SOKOLOV, N.L

PHASE I BOOK EXPLOITATION

sov/1437

Spravochnik metallista v pyati tomakh, t. 4, (Metals Engineering Handbook in Five Volumes, Vol 4) Moscow, Mashgiz, 1958. 778 p. 50,000 copies printed.

- Ed. (Title page): A.N. Malov, Candidate of Technical Sciences; Ed. (Inside book): V.I. Krylov, Engineer; Tech. Ed.: T.F. Sokolova; Editorial Board: N.S. Acherkan (Chairman and Chief Ed.), Doctor of Technical Sciences, Professor; V.S. Vladislawlev, Professor (Deceased); A.N. Malov, Candidate of Technical Sciences; S.N. Pozdnyakov; A. Ya. Rostovykh; G.B. Stolbin; and S.A. Chernavskiy; Managing Ed. for Reference Literature: V.I. Krylov, Engineer.
- This handbook may be useful to technicians and engineers working in the field of machine design and production. PURPOSE:
- COVERAGE: This volume covers the following topics: casting, forging, pressing, stamping, welding, electric methods of machining, and metal cutting, Recently developed electrical methods of machining which are not yet used in production are described; viz., the so-called "electropulse" and "electrohydraulic" methods. No-personalities are mentioned. There are 79 Soviet references.

card. 1/9

Metals Engineering Handbook (Cont.)	sov/1437	
Heating metal before forging and stamping (M.A. Kasenkov, Candid Technical Sciences) General information Cooling of forgings Heating equipment for forging Open die forging(V.B. Pokrovskiy, Engineer) Tools for open die forging Open die forging technique Forging in blacksmith closed dies Combined smith and drop forging by the method of A.V. Potekhi Hummer forging (Yu.V. Shukov, Candidate of Technical Sciences in Hot forging on crank presses (Yu. V.Shukhov, Candidate of Technical Sciences, and N.L. Sokolov, Engineer) Forging on percussion presses Forging on horizontal machines Trimming, piercing, straightening and sizing of forgings (N.L. Sengineer) Trimming and piercing of forgings Straightening of forgings Sizing of forgings	in Lical	62 62 70 73 78 88 93 105 108 109 135 154 155 175 175 175 179 180
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Tools	271
Typical production methods of cold upsetting	272
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Metals Engineering Handbook (Cont.) X. Bench Work (A.W. W. C.) SOV/1437	
Chipping Sawing (A.N. Polyanskiy, Candidate of Technical Sciences) Filing (A.N. Polyanskiy, Candidate of Technical Sciences) Scraping (A.N. Malov, Candidate of Technical Sciences) Leyout (A.N. Malov, Candidate of Technical Sciences) Mechanic's hand tools for assembling (A.N. Malov, Candidate of Technical Sciences) XI. Metal Shearing (N.S. Degtyarenko, Candidate of Technical Sciences) General information	670 670 673 674 700 704 707
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Card 9/9 GO/gmp 5-25-59	

AUTHOR: Sokolov, N.L., Engineer SOV/122-58-6-15/37

FOR THE BUILDING CONTROL OF THE SECRETARISE CONTROL CONTROL OF THE SECRETARISE CONTROL CONTROL

TITLE: The Hot Extrusion Forming of the King Pin of a Motor-car

Front Suspension (Goryachaya shtampovka vydavlivaniyem

stoyki peredney podveski avtomobilya)

PERICDICAL: Vestnik Mashinostroyeniya, 1958, Nr 6, pp 42 - 44(USSR)

ABSTRACT: The hot extrusion of the king pin component weighing

2.8 kg in 0.4% carbon 1% chromium steel is described. Drawings of the component and cross-sections of the die inserts and the die bodies are included. A 4 000-ton crankpress operating at 50 strokes/min is used to press the components out of blanks of 70 mm dia. and 150 mm length and 4.5 kg weight after high-frequency induction heating. The process consists of press extrusion in a

closed die, followed by stamping in an open die. There are 5 figures.

Card 1/1 1. Metals--Extrusion

SOKOLOV, Nikolay Leonidovich, inzh.; KAMNEV, P.V., kand. tekhn. nauk, dots., red.; SLITSKAYA, I.M., red.; FOMICHEV, A.G., red. izd-va; BELOGUROVA, I.A. tekhn. red.

[Modern processes of hot stamping on forging and stamping crank prsses; survey of advaned practice in the automobile industry] Progressivnye protsessy goriachei shtampovki na krivoshipnykh kovochnoshtampovochnykh pressakh; obzor peredovogo opyta avtomobil'noi promyshlennosti. Pod red. P.V.Kamneva. Leningrad. Pt.1. 1961. 52 p. Pt.2. 1961. 55 p. (MIRA 14:7)

(Sheet-metal work)

MOROZOV, A.P., kand. tekhn. nguk; SOKOLOV, N.L., inzh., retsenzent;

[Making dies for forging] Izgotovlenie shtampov dlia goriachei shtampovki. Moskva, Mashinostroenie, 1965. 183 p.

(MIRA 18:5)

SOKOLOV, N. L.

Izuchenie povedeniia zhivotnykh v shkol'nom kurse biologii. (Opyty i nabliudeniia). (Studying the behavior of animals in the school course in biology(experiments and observations)). Moskva, Akad. ped. nauk RSFSR, 1953. 52 p. (Akad. ped. nauk RSFSR. Leningr. in-t pedagogiki. Zaoch. metod. konsul'tatsiia)

SO: Monthly List of Russian Accessions, Vol 7, No. 6, Sep. 1954

REYZIN, Eduard Karlovich; SOKOLOV, N.L., redaktor; PROFERANSOVA, N.V., redaktor; VOLKOV, A.P., tekhnicheskiy redaktor

[Student assignments in beekeeping] Zaniatiia uchashchikhsia po pchelovodstvu. Pod red. N.L.Sokolova. Moskva, Izd-vo Akademii pedagog. anuk RSFSR, 1956. 35 p. (MLRA 9:11)

(Bee culture--Study and teaching)

THE PROPERTY OF THE PROPERTY O

SOKOLOV, Nikolay Leonidovich, kandidat biologicheskikh nauk; YELAGIN, V.D., redaktor; VOLKOV, A.P., tekhnicheskiy redaktor

[Birds; data for the study of bird behavior] Ptitsy; materialy k izucheniiu povedeniia ptits. Moskva, Izd-vo Akademii pedagog. nauk RSFSR. 1956. 82 p. (MLRA 9:11) (Birds)

SOKOLOV, Mikolay Leonidovich; PROFERANSOVA, M.V., red.; LAUT, V.G., tekhn.red.

[Experiments with animals in the schools nature nook] Onyt nad zhivotnymi v ugolke zhivoi prirody. Moskva, Izd-vo Akad.pedagog. nauk HSFSR, 1957. 77 p. (MIRA 11:1)

(Zoology-Study and teaching)

Selection: "Use of the Cast Concrete Piles Puilt in Flace During Underpinning." Descriptions of the Labor Rei Bonner Construction Encineering Instiment V. V. Kuybyshev,

16 Jun 47.

SC: Vechernyaya Mockya, Jun, 1947 (Project #17896)

SERCIOV. and SHEEREST, S.A.

"Primenemiye metallicheskojo Shpunta Priustroystve
di krotekunicheskikh Sooruzheniy"
... mashstrouizdat 1949 160 pp.

The pile-iniver oremain	r; textbook	loskva, 17	skstroijzda:	t, 1949.	183 p.	(62-23325)	
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SOKOLCV, N. M.

Use of grooved steel piling in building deep foundations for forge hammers. Stroi.prom. 29 Foundations No. 12 1951.

1954, Uncl. 9. Monthly List of Russian Accessions, Library of Congress, August

SOKOLOV, N.M., kandidat tekhnicheskikh nauk, nauchnyy redaktor.

[Reinforced concrete construction of skyscraper foundations]
Zhelezobetonnye raboty po ustroistvu fundamentov vysotnykh
Zdanii. Moskva, Gos.izd-vo lit-ry po stroitel'stvu i arkhitekture,
(MLRA 7:3)
1953. 85 p.

1. Moscow. Vsesoyuznyy nauchno-issledovatel'skiy institut organizatsii i mekhanizatsii stroitel'stva.

(Reinforced concrete construction)

(Foundations)

SHVETS, V.B., mladshiy nauchnyy sotrudnik; SOKOLOV, N.M., kandidat tekhnicheskikh nauk, redaktor; PETROVA, V.V., redaktor izdatel stva; MEL'NI-CHENKO, F.P., tekhnicheskiy redaktor

[Instructions for surface compaction of soils for building and structure foundations by means of heavy tamping machines] Ukazaniia po poverkhnostnomu uplotneniiu gruntov v osnovanii zdanii i sooruzhenii tiazhelymi trambovkami. U 136-55/Minstroi. Moskva, Gos. izdvo lit-ry po stroit. i arkhitekture, 1955. 15 p. (MIRA 9:10)

1. Russia(1923- U.S.S.R.) Ministerstvo stroitel'stva.
Tekhnicheskoye upravleniye. 2. Vsesoyuznyy nauchno-issledovatel'skiy institut osnovaniy i fundamentov (for Shvets)
(Soil stabilization)

SOKOLOV, N.M., kandidat tekhnicheskikh nauk, redaktor; AZRILYANT, Ya.M., redaktor izdatel stva; VOLKOV, V.S., tekhnicheskiy redaktor

[Engineering instructions for construction and installation work] Tekhnicheskie usloviia na proizvodstvao i priemku stroitel-nykh i montazhnykh rabot. Moskva, Gos. izd-vo lit-ry po stroit. i nykh i montazhnykh rabot. Moskva, Gos. izd-vo lit-ry po stroit. i arkhitekture. Section 10. [Special foundations] Ustroistvo spearkhitekture. Section 10. [Special foundations] (MIRA 9:9) tsal'nykh osnovanii. 1955. 90 p.

1. Russia (1923- U.S.S.R.) Gosudarstvennyy komitet po delam stroitel'stva.
(Foundations)

SOMOLOV, Nikolay Mikhaylovich, kandidat tekhnicheskikh nauk; SHALYAPIN, R.S., kandidat tekhnicheskikh nauk, redaktor; POLIVANOV, S.I., redaktor izdatel stva; GUSEVA, S.S., tekhnicheskiy redaktor

[Manual on the preparation of rammed concrete piling] Rukovodstva po izgotovleniiu nabivnykh betonnykh chastotrambovannykh svai.

Moskva, Gos. izd-vo lit-ry po stroit. i arkhitekture, 1956. 46 p.

(MIRA 9:10)

KURINK, N.M., kandidat tekhnicheskikh nauk; SOKOLOW, N.M., kandidat tekhnicheskikh nauk; KOPCHUGOV, V.A., kandidat tekhnicheskikh nauk; SOROCHAN, Ye.A., ZAMORIN, P.K., kandidat tekhnicheskikh nauk; SOROCHAN, Ye.A., inzhener; GAROWIIKOV, V.I., inzhener, nauchnyy redaktor; EEGAK, B.A., inzhener; GAROWIIKOV, V.I., inzhener, nauchnyy redaktor; redaktor redaktor izdatelistva; GUSEVA, S.S., tekhnicheskiy redaktor redaktor izdatelistva; GUSEVA, S.S., tekhnicheskiy redaktor sebornykh fundamentov v stroitelistve zdanii. Moskva, Gos. izd-vo (NIRA 10:1) lit-ry po stroit. i arkhitekture, 1956. 77 p. (WIRA 10:1)

SONOLOV, N.M., kandidat tekhnicheskikh nauk.

Building bridge supports across the Yangtze River. Nov. tekh.1

Building bridge supports across the Yangtze River. Nov. tekh.1

pered.op.v stroi. 18 no.6:26-29 Je '56.

(Yangtze River-Bridge construction)

SOROCHAN, Yo.A., kand.tekhn.nauk; SOKOLOV, N.M., kand.tekhn.nauk. Prinimali uchastiye: SEREBRYANYY, H.V.: POL'SHIN, D.Ye., kand.tekhn.nauk. MUNITS, A.P., red.izd-va; BOROVNEV, N.K., tekhn.red.

> [Instructions for using precast footings] Ukazaniia po primeneniiu sbornykh lentochnykh fundamentov. Moskva, Gos. izd-vo lit-ry po stroit., arkhit. i stroit.materialam, 1959. (MIRA 12:10) 28 p.

> 1. Russia (1923- U.S.S.R.) Gosudarstvennyy komitet po delam stroitel stva. 2. Nauchno-issledovatel skiy institut osnovaniy i podzemnykh sooruzheniy Akademii stroitel stva i arkhitektury SSSR (for Sorochan, Sokolov).
> (Foundations) (Pre-

(Precast concrete construction)

TSYTOVICH, N.A., prof.; VESELOV, V.A., dotsent, kand.tekhn.nauk; KUZ'MIN, P.G., dotsent, kand.tekhn.nauk; FERROHSKIY, V.I., kand.tekhn. nauk, assistent; PILYUGIN, A.I., kand.tekhn.nauk, assistent; LUGA, A.A., kand.tekhn.nauk, starshiy nauchnyy sotrudnik; SOKO-LOV, N.M., kand.tekhn.nauk, starshiy nauchnyy sotrudnik; SAVINOV, O.A., doktor tekhn.nauk; KOSTERIN, E.V., kand.tekhn.nauk, assistent. Prinimali uchastiye: AKINSHIN, V.M.; MARTSENYUK, V.I., starshiy laborant. VASIL'YEV, B.D., prof., doktor tekhn.nauk, retsenzent; BEREZANTSEV, V.G., prof., doktor tekhn.nauk, retsenzent; LAGAR'KOV, N.I., inzh., nauchnyy red.; SMIRNOVA, A.P., red.izd-va; NAUMOVA, G.D., tekhn.red.

[Foundation engineering] Osnovaniia i fundamenty. Pod red. N.A. TSytovicha. Moskva, Gos.izd-vo lit-ry po stroit., arkhit. i (MIRA 13:5) stroit.materialam, 1959. 452 p.

1. Chlen-korrespondent AN SSSR (for TSytovich). 2. Zaveduyushchiy laboratoriyey kafedry osnovanjy i fundamentov Moskovskogo inzhenerno-stroitel nogo instituta imeni V.V.Kuybysheva (for Akinshin). 3. Zaveduyushchiy kafedroy osnovaniy 1 fundamentov Leningradskogo instituta inzhenerov zheleznodorozhnogo transporta imeni akademika V.N.Obraztsova (for Berezantsev).

(Soil mechanics)

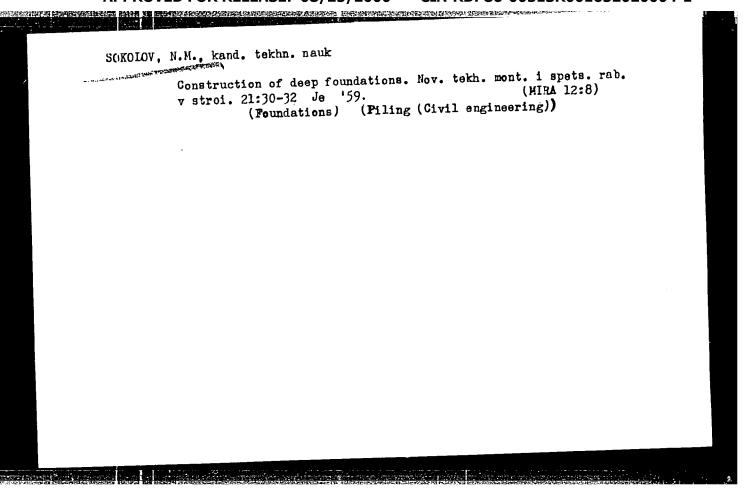
(Foundations)

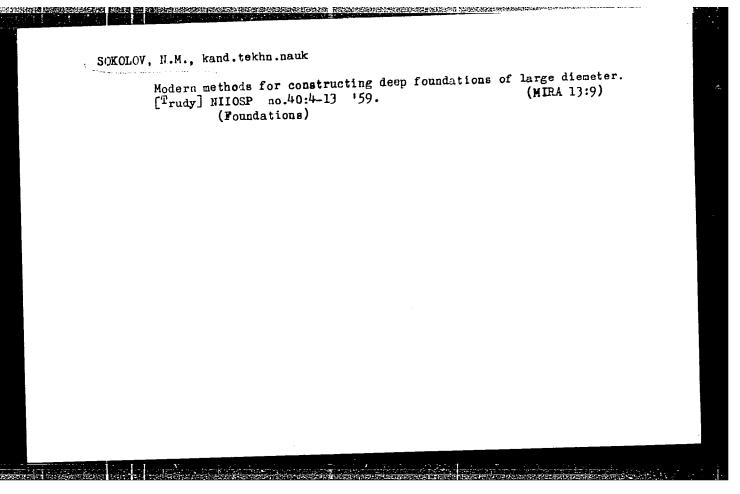
SOKOLOV, N.M.; TOKAR', R.A.

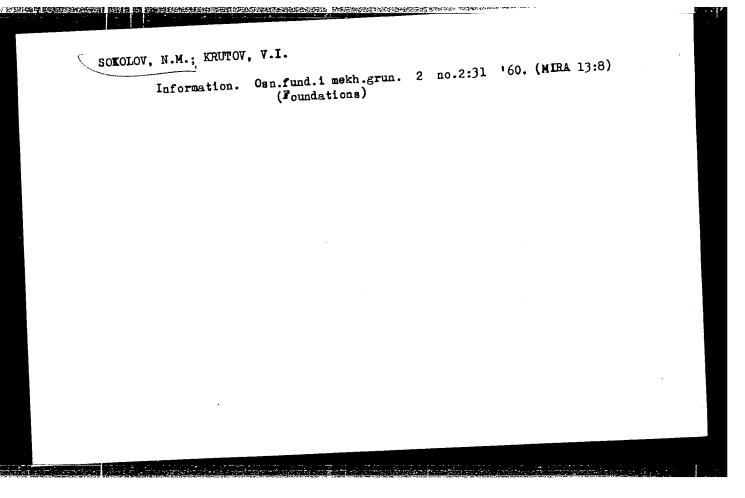
Industrial methods of constructing foundation beds and foundations.

Osn., fund. i mekh. grun. no.2:1-5 '59. (MIRA 12:7)

(Foundations) (Soil stabilization)







Total Control of the		
sokolov, M. I studies, ele methods." M (KL, 8-61, 2	No., Doc Tech Sci "Report on the tasks of foundation-building contains, and introduction of foundation-building contains, and foundation full os, 1961. (Acad of Build and Architec USSR)	
	-188 - - 188 -	

BOKUYANOV, A.I., inzh., red.; SOKOLOV, N.M., doktor tekhn. nauk, red; TIKUMOV, P.R., kend. tekhnicheskikh nauk, red.; STRASHNYKH, V.P., red. 12d-va; NAUMOVA, G.D., tekhn. red.

[Construction specifications and regulations] Stroitel'nye normy i pravila. Moskva, Gosstroiizdat. Pt.3. Sec.B. ch.6. [Foundations and supports of piling and shells, pile walls; regulations for production and acceptance of work] Fundamenty i opory iz svai i obolochek, shpuntovye ograzhdeniia; pravila proizvodstva i priemki rabot (SNIP III-B. 6-62). [MIRA 16:9]

1. Russia (1923— U.S.S.R.) Gosudarstvennyy komitet po delam stroitel'stva. 2. Gosudarstvennyy komitet Soveta Ministrov SSSR po delam stroitel'stva (for Bokunyayev). Mezhduvedom-stvennaya komissiya po peresmotru Stroitel'nykh norm i pravil Akademii stroitel'stva i arkhitektury SSSR (for Sokolov).

3. Nauchno-issledovatel'skiy institut osnovaniy Akademii stroitel'stva i arkhitektury SSSR (for Tikunov).

(Piling (Civil engineering)) (Foundations)

SOKOLOV, N.M., doktor tekhn. nauk; MARIUPOL'SKIY, G.M., kand. tekhn. nauk; SOROCHAN, Ye.A., kand. tekhn. nauk; FUKSON, M.N., kand. tekhn. nauk; YEFREMOV, M.G., kand. tekhn. nauk, retsenzent; ZHURAVLEV, B.A., red.; SHEVCHENKO, T.N., tekhn. red.

[Foundation engineering] Osnovaniia i fundamenty. [By] N.M. Sokolov i dr. Moskva, Gosstroiizdat, 1963. 295 p.

(MIRA 16:12)

(Foundations)

SOSHIN, A.V., doktor tekhn. nauk, prof. SOKOLOV, N.M., doktor tekhn. nauk, prof.; TCHOFOL, A.S., kand. tekhn. nauk, dots.; BELINOVICH, M.S., inzh.; PETROV, N.S., kand. tekhn. nauk; LUPENKO, I.S., inzh., nauchn. red.

[Technology of the construction industry] Tekhnologiia stroitel'nogo proizvodstva. [By] A.V.Soshin i dr. Moskva, stroitedat, 1964. 423 p. (MIRA 17:10)

ABELEV, Yu.M., doktor tekhm. nauk, prof.; ABELEV, M.Yu., inzh.;

BAKHOLDIN, B.V., kand. tekhn. nauk; BEREZANTSEV, V.G.,

doktor tekhm. nauk, prof.; VYALOV, S.S., doktor tekhn.

nauk; GODES, E.G., inzh.; GORBUNOV-POSADOV, M.I., doktor

tekhm. nauk, prof.; DAINATOV, B.I., doktor tekhm. nauk,

prof.; DOKUCHAYEV, V.V., kand. tekhm. nauk; KRUTOV, V.I.,

kand. tekhm. nauk; KSENOFONTOV, A.I., kand. tekhm. nauk;

MARIUPOL'SKIY, G.M., kand. tekhm. nauk; MORARESKUL, N.N.,

inzh.; PERLEY, Ye.M., inzh.; SAVINOV, O.A., doktor tekhm.

nauk; SIDOROV, N.N., kand. tekhm. nauk; SMORODINSKIY,

N.I., kand. tekhm. nauk; SOKOLOV, N.M., doktor tekhm.nauk;

EYEKOV, M.L., inzh.; SHASHKOV, S.A., kand. tekhm.nauk,

[GECEASEd]; KHALIZEV, Ye.P., kand. tekhm. nauk, nauchm.red.

[Manual for the designing of industrial plants, apartment houses, and public buildings and structures; foundations] Spravochnik proektirovshchika promyshlennykh, zhilykh i obshchestvennykh zdanii i sooruzhenii; osnovaniia i fundamenty. Leningrad, Stroiizdat, 1964. 268 p. (MIRA 18:1)

SOKOLOV, Nikolay Mikhaylevich, doktor tekhn. nauk; KHUTOV, Vladimir Ivanovich, kand. tekhn. nauk; SOHOCHAN, Yevgeniy Andreyevich, kand. tekhn. nauk;

[Construction of large-panel buildings on sagging ground] Stroitel'stvo krupnopanel'nykh zdanii na prosadochnykh gruntakh. Moskva, Stroiizdat, 1965. 191 p.

(MIRA 18:2)

SOKOLOV, Nikolay Mikhaylovich; MIRKIN, M.S., red.

[Using statistical control methods in introducing a flaw-less production system] Opyt ispol'zovaniia statistiche-skikh metodov kontrolia pri vnedrenii sistemy bezdefektnogo izgotovleniia produktsii. Leningrad, 1965. 26 p.

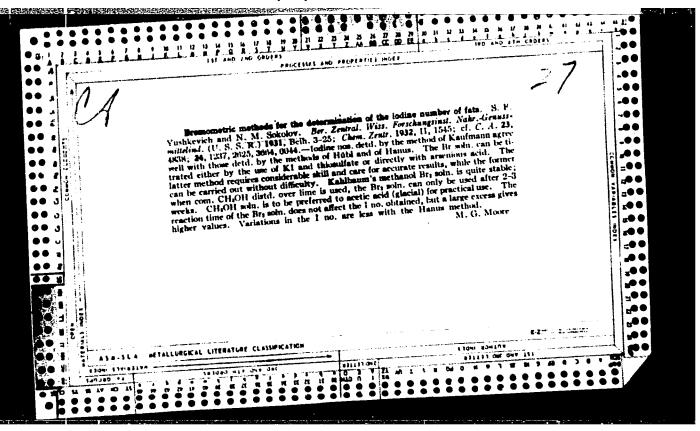
(MIRA 18:7)

SOKOLOV, N.M.; SEVRIUGOVA, N.N.; ZHAVORONKOV, N.M.

Liquid-vapor equilibrium in the system acrylonitrile-acrolein at various pressures. Zhur. fiz. khim. 39 no.4:1008-1012 Ap '65.

(MIRA 19:1)

1. Institut obshchey i neorganicheskoy khimii AN SSSR. Submitted May 27, 1964.



SOKOLOV, N. M.

USSR/ Chemistry

Physical chemistry

Card

: 1/1

Pub. 151 - 11/35

Authors

: Sokolov, N. H.

Title

: Investigation of the liquidus curve in binary systems formed by scdium

salts of fatty acids with sodium nitrate and rhodanide

Periodical.

: Zhur. ob. khim. 24, Ed. 7, 1150 - 1156, July 1954

Abstract

Liquidus curves of seventeen binary systems, composed of sodium salts of fatty acids and NaNO3 and NaSCN, were investigated by the visual polythermal analysis method. Data on the homology and isomerism of fatty acid salts, are presented. The reasons for the changes in the chemical properties of fatty acid salts, which take place during the change in the number of C-atoms in the radical and during isomerization of the latter, are outlined. Thirteen references: 5 USSR, 5 German and

3 Italian. Tables, graphs.

Institution

: Medical Institute, Smolensk

Submitted

: February 9, 1954

CIA-RDP86-00513R001652020004-1 "APPROVED FOR RELEASE: 08/25/2000

SOKOLOV, NM.

USSR/Chemistry

Pub. 151 - 20/42 1/1 Card

Sokolov, N. M. Authors

Reaction of sodium salts of carboxylic acids in fusions Title

Zhur. ob. khim. 24/9, 1581-1593, Sep 1954 Periodical

The reaction of binary systems composed of sodium salts of carboxylic acid in fusions was investigated by the visual polythermal method. Abstract

Certain laws pertaining to the effect produced by the paraffinic part of the molecule of the carboxylic acid salt on the processes occurring in salt fusions were established. The results obtained from the study of 28 binary systems, formed by sodium salts of carboxylic acids, are described. Nine references: 3-USSR; 5-German and 1-USA (1882-1947).

Tables; graphs.

Institution : Medical Institute, Smolensk

: April 14, 1954 Submitted

USSR/Physical Chemistry - Thermodynamics, Thermochemistry,

B-8

Equilibria, Physical-Chemical Analysis, Phase Transitions.

Abs Jour : Referat Zhur - Khimiya, No 1, 1958, 413

Author : N.M. Sokolov.

Inst : Title : Interaction of Sodium Salts of Aliphatic Acids with Sodium

Nitrites in Melts.

Orig Pub : Zh. obshch. khimii, 1957, 27, No 4, 840-844

Abstract : 8 binary systems of salts of aliphatic acids containing

various radicals and of NaNO₂ were studied by the visual-polythermal method. The capacity to produce complexes (systems with propinate and butyrate) appears with the rise of the number of carbon atoms in the radical, but at a further rise of the number of carbon atoms, this capacity disappears (system with valerate) and still further the salts looses the capacity to mix with the nitrite (system with capronate). Isomer radicals in salts do not promote

Card 1,/2

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LUTHORS :

Scholov, N. M., Pochtakova, E. I.

79-28-5-64/69

TITLE:

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Reciprocally Irreversible System of Formatos and

Throughrates of Sodium and Potassium (Neobratimovzaimnaya sistema iz formiatov i rodanidov natriya i

kaliya)

PERIODICAL:

Zhurnal Obshchey Khimii, 1958 Vol 28, Nr 5,

pp. 1391-1397 (USSR)

ABSTRACT:

The investigation of the mutual systems of the salts of fatty acids and thiocyanates of sodium and/ out in order to determine the influence of the structure of the radical of the salt of a fatty acid on the melting processes of the salts. As one of the lateral lines in the phase diagram of the systems to be investigated served the binary system NaCNS KCNS; the sodium and potassium salts of the homologous series of fatty acids with straight as well as with ramified chains of carbon atoms in the radical were taken as the other salts. In the present work the mutual system Na, K | HCOO, CNS was investigated. The sodium thicovanate does not show polymorphous conversions

card 1/3

Reciprocally Trreversible System of Formares and Thiocya 79-28-5-64/69 nates of Sodium and Porassium

and remains unchanged up to 50010 (References 3,2). The potassium thiocyanate is thermally resistant up to 400°C and at 142.5°C suffers a polymorphous conversion (Reference rence 3). The sodium formate retains this property till 310°C (Reference 4), the polymorphous conversion (Reference 5) beginning at 242°C. At 370°C the potassium formate begins to decompose; at 470°C the decomposition takes place very quickly (Reference 6). On the heating of po rassium formate by means of the differential thermocouple the polymorphous conversions for it were found at 60, 135 and 570C (Reference 5). For all components of the gived system the following data of the thermal effect are known (im k hal equ.) 3 NaCNS-40,36 KCNS-47:40 HCOONa-57:7: HCOOK-160.7 (Reference 7) Based on these data it can be expected that the equilibrium in this system is displaced in the direction to the pair HCOONa and KCNS: HCOOK+ NaCNS - HC90Na+KCNS+4.oc k cal. Thus, by means of the visual-thermal method the binary system KCNS-HCOOK and the mutual system of formates and thioryanates of potas-

Card 2/3

Irreversible System of Formates and Thiocya- 79-28-5-64/69 nates of Sodium and Potassium

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sium and sodium were investigated for the first time. There are 6 figures, 2 tables and 9 references, 6 of which are Soviet.

ASSOCIATION: Smolenskiy meditsinskiy institut (Smolensk Medical

Institute)

SUBMITTED: March 5, 1957

Card 3/3

AUTHORS: Sokolov, H. M. Pochtakova, E. I. 79-28-5-65/69

Reciprocally Irreversible System of Acetates and Propio-TITLE: nates of Sodium and Potassium (Neobratimo-vzaimnaya siste-

ma iz atsetatov i propionatov natriya i kaliya)

PERIODICAL: Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr. 5,

pp 1397-1404 (USSR)

The direction of conversions in of sodium and potas-ABSTRACT: sium salts of organic acids has been little investigated hitherto. Among many reciprocalstems mentioned in references

there are none consisting of organic acids. The present system is the first link in the chair of systems being

used to determine the relation between the

reaction direction and the structure of the salt radical. Among the investigated components of the system,

sium acetate was of prime consideration, while calcium acetate and the propionates were secondary. The merting point of acetate acetate is between 319 and 33700 (References 1-7); that of potassium acetate at 292-3020 (References 2,4,8000); that

Card 1/3

of potassium proprionate above 30000 (References 11), All

The Mutually Irreversible System of Acetates and Proprior 79-28-5-65/69 nates of Sodium and Potassium

THE PROPERTY OF THE PROPERTY O

salts melt without decomposition and are thermally unchangeable up to 3900C. Vorlender (Reference 12) described the polymorphous conversions of potassium and sodium proprionate, however, - not of the acetates of these metals. Also A. Baskov (Reference 2), by means of the differentialthermal method of analysis, investigated the binary system of potassium and sodium acetates without finding polymorphism of the initial salts. G. G. Diogenov in the investigation of the mutual system of acetates and nitrates of lithium and sodium wrongly reports on the presence of two modifications α and β with respect to sodium acetate at 323°C, as the investigation by means of the differential thermocouple showed that no conversions take place at this temperature. One of the authors found by means of this methed polymerphous conversions of sodium proprionate at 77° , 195°, 217° and 287°C; of potassium proprionate at 330°C. Thus the binary systems: CH3COOK C3H5COOK; CH3COONA C3H5 COONa, and $C_2H_5\text{COONa}$, $C_2H_5\text{COOK}$ were investigated for the acetates of sodium first time. The binary system of

Card 2/3

The Reciprocally Irreversible System of Acetates and 79-28-5-65/69

Propionates of Sodium and Potassium

and potassium in which complex formation was established was repeated. The equilibrium in the system is displaced in the direction to the highest melting components. There are 8 figures, 2 tables and 13 references, 5 of

which are Soviet.

ASSOCIATION:

Smolenskiy meditsinskiy institut (Smolensk Medical

Institute)

SUBMITTED:

January 25, 1957

Card 3/3

AUTHORS:

Tsindrik, N. M., Sokolov, N. L.

表现 **医线线 医聚性结形性切迹**医结束处理的 化对抗原则处于检查点 不过原性的 的复数次次原则从正常是在上面的企业。

79-28-5-66/69

TITLE:

The Triple Recipical System of Propionates and Nitrates of Lithium and Sodium (Troynaya vzaimnaya sistema iz propionatov

i nitratov litiya i natriya)

PERIODICAL:

Zhurnal Obshchey Khimii, 1958, Vol. 28, Nr 5,

pp. 1401 - 1410 (USSR)

ABSTRACT:

The given system is one of a series of systems being investigated to establish the dependence of the direction of a reaction (between the lithium salt of an organic acid and sodium nitrate) on the length of the carbon chain of an acid radical. Earlier it was shown that in similar cases the equilibrium in formates and acetates is displaced in the direction to sodium nitrate and lithium salt of the fatty acid (References 1,2). In the acetate system the equilibrium is more displaced. The given system must explain whether the character of the displacement is subject

to rules. Different from nitrates, the lithium- and sodiumproprionates are little investigated. Thus, their melting points were determined for the first time only recently

card 1/3

79-28-5-66/69

The Triple Reciprocal System of Propionates and Nitrates of Lithium and Sodium

(Reference 4). Vorlender (Forlender), without giving temperature data is of the opinion that sodium propionates has polymorphous conversions. One of the authors found polymorphous conversions in sodium propionate at 77,195,217 and 287°C (Reference 3). All salts of the present system melt without decomposition. On overheating, the propionates darker and decompose forming a gas. The mixture of nitrates and propionates is on this treatment accompanied by a flashing. Thus the two double systems LiNO3-C2H5COOLi and C2H5COONa-C2H5COOLi were investigated. The melting diagram of the triple redrocal system of propionates and nitrates of potassium and sodium was set up. The equilibrium in the triple system is displaced in the direction to the lithium propionate and sodium nitrate; when the carbon atoms are increased in the fatty acid radical, the displacement of equilibrium increases in systems composed of formates, acetates

Card 2/3

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79-28-5-66/69

Reciprocal
The Triple System of Propionates and Mitrates of Lithium and Sodium

or propionates with nitrates of lithium and sodium. There are 8 figures, 3 tables and 6 references, 5 of which are Soviet.

ASSOCIATION: Smolenskiy meditsinskiy institut (Smolensk Medical Institute)

SUBMITTED: March 27, 1957

Card 5/3

CIA-RDP86-00513R001652020004-1 "APPROVED FOR RELEASE: 08/25/2000

507/79-28-6-57/63 Sokolov, N. M., Pochtakova, E. I.

Reversible Mutual System of Butyrates and Sodium-and : EFOHTUA:

Potassiumthiocyanates (Neobratimo-vzaimnaya sistema iz TITLE:

butiratov i rodanidov natriya i kaliya)

Zhurnal obshchey khimii, 1958, Vol. 28, Nr 6, pp.

PERIODICAL: 1693-1700 (บรรส)

The main task of the present paper was the investigation of the conversion process in the alloys of butyrates and ABSTRACT:

thiocyanates of sodium and potassium. The given mutual system

represents a certain part in the series of salts of

thiocyanic acid and of fatty acids, the common aim of investigation consists in the determination of the dependence of the direction of reaction on the structure of the radical of the fatty acid (Ref 1). Potasium thiocyanate is subjected

to polymorphous conversion at 1430 (Ref 2). Sodium thiocyanate has, according to data in publications (Ref 2), no polymorphous

conversion. The investigation carried out by one of the

authors (Ref 3) showed that polymorphism occurs in butyrates.

Thus sodiumbutyrate suffers a conversion at 117,232, 252 and

Card 1/3

Reversible Mutual System of Butyrates and Sodium- SOV/ 79-28-6-57/63 -and Potassivethiocyanates

The whole investigation was carried out according to the visual-polythermal method as generally used. All compositions visual-polythermal method as generally used. All compositions in molar per cents, together with formation temperatures of in molar per cents, together with formation temperatures of in molar per cents, together with formation temperatures of in molar per cents, together with formation temperatures of in molar per cents, together with formation temperatures of in molar per cents, together with a liquidus the first crystals are given (Tables 1 - 3). The liquidus curves (likvidus) of the double system C₃H₇COONa-C₃H₇COOK and KCNS-C₃H₇COOK were plotted. With an increase of the number of carbon atoms in the salt radical also the degree of reversibility in such systems increases which had been of reversibility in such systems increases which had been formed of the fatty acid salt of sodium- and potassium thiocyanates. There are 6 figures, 4 tables, and 6 references, 6 of which are Soviet.

ASSOCIATION:

Smolenskiy meditsinskiy institut (Smolensk Medical Institute)

SUBMITTED:

April 3, 1957

Card 2/3

Reversible Mutual System of Butyrates and Sodium- $\frac{507}{9-28-6-57/63}$ -and Potassiumthiocyanates

1. Sodium thiocyanates--Chemical reactions 2. Potassium thiocyanates--Chemical reactions

Card 3/3

AUTHORS:

Tsindrik, N. M., Sokolov, N. M.

SOV/79-28-7-3/64

TITLE:

Ternary Mutual System of Butyrates and Nitrates of Lithium and Sodium (Troynaya vzaimnaya sistema iz butiratov i nitratov

litiya i natriya)

PERIODICAL:

Zhurnal obshchey khimii, 1958, Vol. 28, Nr 7,

pp. 1728 - 1733 (USSR)

ABSTRACT:

The present investigation was carried out to determine the direction of the conversion reaction in melts of butyrates and nitrates of lithium and sodium. It is of interest to compare this system to the earlier investigated ones of formiates, acetates, propionates of lithium and sodium, and the nitrates of the same metals. The surface of the molten systems in a still liquid state was investigated according to the visual-polythermal method. A nickel-chromium thermocouple and a millivoltmeter were used for the determination of the temperature, where the first crystals appeared. The butyrates of lithium and sodium were obtained by the addition of excess butyric acid to their carbonates and by a subsequent evaporation (Ref 1). The dry salts obtained were purified by recrystallization of butanol. The

Card 1/3

Ternary Mutual System of Butyrates and Nitrates of SOV/79-28-7-3/64 Lithium and Sodium

melting points of the system components were: LiNO₃-256°, NaNO₃-308°, C₃H₇COOLi-329°, C₃H₇COONa-330°. With NaNO₃ a polymorphous transformation was found at 275°, with C₆H₇COONa at 117, 232, 252 and 316° (Ref 2). The investigation of the molten ternary mutual system of butyrates and nitrates of lithium and sodium is new. The data on the two double systems LiNO₃-C₃H₇COOLi and C₃H₇COOLi-C₃H₇COONa are described for the first time (Ref 7). The dependence of the chemical dislocation of equilibrium in the series of the systems of formiates, acetates, propionates and butyrates with nitrates of lithium and sodium on the number of carbon atoms in the radical of the fatty acid was determined. There are 7 figures, 3 tables, and 5 references, 5 of which are Soviet.

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Card 2/3

Ternary Mutual System of Butyrates and Nitrates of

SOV/79-28-7-3/64

Lithium and Sodium

ASSOCIATION:

Smolenskiy gosudarstvennyy meditsinskiy institut (Smolensk

State Medical Institute)

SUBMITTED:

September 6, 1957

1. Lithium nitrates—Chemical reactions 2. Sodium nitrates
Chemical reactions 3. Lithium butyrates—Chemical reactions
4. Sodium butyrates—Chemical reactions 5. Metalorganic

compounds--Properties

Card 3/3

CIA-RDP86-00513R001652020004-1 "APPROVED FOR RELEASE: 08/25/2000

AUTHORS:

SOV/79-28-11-3/55 Dmitrevskaya, O. I., Sokolov, N. M.

TITLE:

Ternary Reciprocal System of Propionates and Nitrates of Sodium and Potassium (Troynaya vzaimnaya sistema iz

propionatov i nitratov natriya i kaliya)

PERIODICAL:

Zhurnal obshchey khimii, 1958, Vol 28, Nr 11, pp 2920-2926

(USSR)

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ABSTRACT:

In the earlier investigated systems of formiates and nitrates of sodium and potassium (Ref 1) as well as in the system of acetates and nitrates of the same metals (Refs 2,3) the direction of the reaction agrees with the conditional thermochemical effect of the reaction, with the degree of irreversibility in the acetate system being higher than in the formiate system. The present system is a further member in the series of these systems. The propionates are little investigated as compared to formiates and acetates. The determination of their heat of formation (Ref 4) is lacking. Therefore, the comparison of the results of the present system can be carried out only with respect to the quantity of the paraffin part in the melts of the process taking place. The melting points and the presence of the polymorphic

Card 1/3

Ternary Reciprocal System of Propionates and Nitrates SOV/79-28-11-3/55 of Sodium and Potassium

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transformations for the nitrates were determined by several scientists; the same data were obtained also for propionates by one of the authors. The ternary reciprocal system Na,K \parallel C₂H₅COO,NO₃ as well as for the first time the double system $C_2H_5COOK-KNO_3$ were investigated. The reaction of the reaction cleavage in the reciprocal system dominates over the complex formation. The formation of solid solutions on the stable diagonal line C2H5COONa-KNO3 was found, which fact is opposed to the theory of isomerism, and demands further investigations. The system Na, K | C2H5COO, NO3 is classified as an irreversible, reciprocal system with a stable diagonal line. The influence of the atomic number of carbon on the degree of irreversibility of the reaction between the potassium salt and fatty acid and sodium nitrate was explained. There are 8 figures, 2 tables, and 17 references, 12 of which are Soviet.

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Ternary Reciprocal System of Propionates and Nitrates SOV/79-28-11-3/55 of Sodium and Potassium

ASSOCIATION: Smolenskiy gosudarstvennyy meditsinskiy institut (Smolensk Medical State Institute)

SUBMITTED: September 5, 1957

Card 3/3

KAZANSKIY, B.A.; LUKINA, N.Yu.; NAKHAPETYAN, L.A.; ZOTOVA, S.V.;
LOZA, G.V.; SHATENSHTEYN, G.A.; OVODOVA, V.A.; UVAROV, O.V.;
SOKOLOV, N.M.; SKOL'NIKOV, V.P.

Production of high purity cyclopropane. Knim. prom. no. 6:462-465 S '60.

(Cyclopropane)

5.4200

77343 807/79-30-1-4/78

AUTHORS:

Dmitriyevskaya, O. I., Sokolov, N. M.

TITLE:

Ternary Interacting System of Sodium- and Potassium

Isobutyrates and Nitrates

PERIODICAL:

Zhurnal obshchey khimii, 1960, Vol 30, Nr 1, pp 20-

25 (USSR)

ABSTRACT:

The known experimental data on various ternary interacting systems leave an impression that heteroipnic compounds occur when an aliphatic acid with a branched carbon chain is one of the components. The other components can be represented by any Na or K salts. The authors seek to substantiate this concept. They carried out experiments using recrystallized and chemically

pure samples of commercial Na and K nitrates and isobutyrates. The mp of NaNO $_3$, KNO $_3$, iso-C $_3$ H $_7$ COONa, and iso-C $_3$ H $_7$ COOK are 308°, 337°, 260°, and 356° C,

Card 1/7

respectively. The first compound undergoes phase

Ternary Interacting System of Sodium- and Potassium Isobutyrates and Nitrates

77343 sov/79-30-1-4/78

transition in solid state at 270°C; the second at 124° and 316°C; the third at 67°C, 91°C, and 220°C, and the fourth at 208°C, 273°C and 348°C. Binary systems NaNO3 - iso-C3H7COONa; iso-C3H7COONa - iso-C3H7COOK; and NaNO3-KNO3 are known. The pair, iso-C3H7COOK - KNO3, studied for the first time, proved to form a heteroinonic compound of iso-C3H7COOK·KNO3 composition when mixed at intermediate ratios (Fig. 1). Heteroionic compounds are also formed in binary systems KNO3 - iso-C3H7COONa and NaNO3- iso-C3H7COOK which represent diagonals in the state diagram (Figs. 3 and 8) of four component systems. Besides the binary systems, 16 ternary sections, denoted in Fig. 3 by Roman numerals, were studied, and their solubility curves and ivariant points were found. The four-component diagram of Fig. 8, based on these data, shows the solid-liquid equilibrium surface, divided into 6 areas, within which 6 solids of different compositions crystallize, 4 mono- and 2 heteroionic.

Card 2/7

Ternary Interacting System of Sodium- and Potassium Isobutyrates and Mitrates

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The central area seems to correspond to $1\text{so-C}_3\text{H}_7\text{COONa}\cdot\text{KNO}_3$ composition. The 10 ternary eutectic curves in the diagram mean equilibrium of 2 solid phases and the liquid. The curves intersect at 4 invariant eutectic points \mathbf{E}_1 , \mathbf{E}_2 , \mathbf{E}_3 , \mathbf{E}_4 and 1 transitional point P at which the following groups of solid phases crystallize, respectively: (1) $1\text{so-C}_3\text{H}_7\text{COONa}\cdot\text{KNO}_3$, $1\text{so-C}_3\text{H}_7\text{COOK}$, and $1\text{so-C}_3\text{H}_7\text{COONa}$ at the ratio $1\text{so-C}_3\text{H}_7\text{COONa}\cdot\text{isc-C}_3\text{H}_7\text{COOK}$: $1\text{co-C}_3\text{H}_7\text{COONa}$ at the ratio $1\text{co-C}_3\text{H}_7\text{COONa}\cdot\text{KNO}_3$, $1\text{co-C}_3\text{H}_7\text{CO$

Card 3/7

Ternary Interacting System of Sodium- and Potassium Isobutyrates and Nitrates

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isc- 0_3H_7 0COK·KNC3 at the ratio isc- 0_3H_7 00ONa:

:iso-0.3H₇GOOK:KNC₃ = 11:44:45%. There are 8 figures;

2 tables; and 10 references, 9 Soviet, 1 U.S. The U.S. reference is: F. C. Krachek, J. Am. Chem. Soc.,

53, 2607, 1931.

ASSOCIATION:

Smolensk State Medical Institute (Smolenskiy gosuđar-

stvennyy meditsinskiy institut)

SUBMITTED:

January 7, 1959

Card 4/7

SOKOLOV, N.M.; POCHTAKOVA, E.I.

Adiagonal reciprocal ternary system of sodium and potassium butyrates and acetates. Zhur.ob.khim. 30 no.5:1401-1405 My '60. (MIRA 13:5)

1. Smolenskiy meditsinskiy institut.
(Butyric acid) (Acetic acid)

ARTHUR BESTELLE BERSCHOOLSTELLE BERSCHOOLSTELLE BERSCHOOLSTELLE BERSCHOOLSTELLE BERSCHOOLSTELLE BERSCHOOLSTELLE

(MIRA 13:5)

SOKOLOV, N.M.; POCHTAKOVA, E.I. Reciprocal ternary system of sodium and potassium isobutyrates and acetates. Zhur.ob.khim. 30 no.5:1405-1410 My '60.

1. Smolenskiy meditsinskiy institut.
(Isobutyric acid) (Acetic acid)

CIA-RDP86-00513R001652020004-1" APPROVED FOR RELEASE: 08/25/2000

SOKOLOV, N.M.; MINICH, M.A.

Liquidus curve in the binary systems formed by molten potassium salts of fatty acids and potassium nitrite. Zhur.neorg.khim. 6 no.11:2558-2562 161. (MIRA 14:10)

1. Smolenskiy gosudarstvennyy meditsinskiy institut, kafedra neorganicheskoy i analiticheskoy khimii.

(Potassium salts) (Systems (Chemistry))

SOKOLOV, N.M.; TSINDRIK, N.M.; DMITREVSKAYA, O.I.

Layering in ternary reciprocal systems consisting of salts of organic and inorgornic acids. Zhur. o.b khim. 31 no.4:1051-1056 Ap '61. (MIRA 14:4)

1. Smolenskiy meditsinskiy institut. (Systems (Chemistry))

"APPROVED FOR RELEASE: 08/25/2000 43783 G/025/62/000/004-5/004/005 1041/1241 Uvarov, O.V., Sokolov, N.M., and Zavosonokov, N.M. 11.3600 Physico-chemical constants of H2018 AUTHORS: PERIODICAL: Kernenergie, no.4-5, 1962, 323-329; TITLE: The elementary separation factor for the system H₂Ol6_

H2018 in the temperature range from 20-210°C was determined by a now in the temperature range from 20-210-6 was determined by a differential vapour pressure measurement method. The results are differential vapour pressure measurement method. From these results given by the formula $\alpha = 0.9835$ exp (7.598/T). From these results one calculates the difference in latent heat of evaporation of the one calculates the difference in latent heat of evaporation of mineral contents. two water species as 14.98 cal/niob and the boiling point of pure H2018 at atmospheric pressures as 100.13°C. The refractive index n_2 at atmospheric pressures as 100.13°C. The refractive index difference — Δn — between light and heavy water was measured at n_2 0°C with the results: n_2 0°C with the results: n_3 0°C was found to be of the refractive index difference between 100 – 30°C was found to

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EVAROV, O.V.; SOKOLOV, N.M.; LYAPIN, V.V.; ZHAVORONKOV, N.M.

Coefficients of separation of the carbon isotopes cl2 - cl4
during the equilibrium vaporization of methane. Zhur. VKHO
7 no.6:695-697 '62. (MIRA 15:12)

1. Nauchno-issledovatel'skiy fiziko-tekhnicheskiy institut
imeni L.Ya. Karpova. (Methane)
(Carbon--Isotopes)
(Evaporation)

43470

S/076/62/036/012/005/014 B101/B180

11 330

Uvarov, O. V., Sokolov, N. M., and Zhavoronkov, N. M. (Moscow)

AUTHORS: Uvarov, U. v., Santa of heavy oxygen water
TITLE: Physical and chemical constants of heavy oxygen water

PERIODICAL: Zhurnal fizicheskoy khimii, v. 36, no. 12, 1962, 2699 - 2706

TEXT: Water containing $\sim 47\%$ H₂0¹⁸ and up to 60% D₂0 was purified, and the D₂0 content was reduced to 0.016 mole% with reduced heated metallic hot iron. The following physicochemical constants were determined: (A) The H₂0¹⁶ - H₂0¹⁸ separation coefficient \approx by a differential method similar to that used by W. H. Keesom, J. Haantjes (Physica, 2, 986, 1935) for separating neon isotopes. Result: between 20 and 210°C, $\log \approx 3.300/T$ separating neon isotopes. Result: between 20 and 210°C, $\log \approx 3.300/T$ separating neon isotopes. Result: between 20 and 210°C, $\log \approx 3.300/T$ separating neon isotopes. Result: between 20 and 210°C, $\log \approx 3.300/T$ separating neon isotopes. The difference in heats of vaporization is 14.97 cal/mole, ≈ 1.0036 . The difference in heats of vaporization is 14.97 cal/mole, the boiling point of H₂0¹⁸ at 760 mm Hg is 100.13°C. (B) The refractive index was determined with an interferometer. $\Delta n = 0.00034N$ H₂018 holds for Card 1/2

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Physical and chemical ...

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white light at 20°C; $N_{\rm H_2O}^{18}$ is the molar part of $H_2^{0}^{18}$ in the mixture. Between 10 and 30°C, the temperature coefficient of the difference in refractive indices of H_2^{016} and H_2^{018} is $(1.18-1,20)\cdot 10^{-6}$. (C) The water density was determined pycnometrically for different H_2^{018} contents. Results: $d_4^t = A + 0.001070 N_{\rm H_2O}^{18}$, where A = 0.99720 at 25° C, 0.99580 at 30° C, and 0.99230 at 40° C. At 25° C, the density of 100% H_2^{018} is 1.10724 with respect to river water. There are 3 figures and 5 tables. The most important English-language references are: S. Sakata a. N. Morita, Bull. Chem. Soc. Japan, 29, 284, 1956; H. E. Watson, J. Amer. Chem. Soc., 76,

ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya. Karpova (Physicochemical Institute imeni L. Ya. Karpov)

SUBMITTED:

June 23, 1961

Card 2/2

SOKOLOV, N.M.; UVAROV, O.V.

Betrachtung einiger Methoden zur Bestimmung der Koeffizienten der relativen Fluchtigkeiten stabiler Isotope.

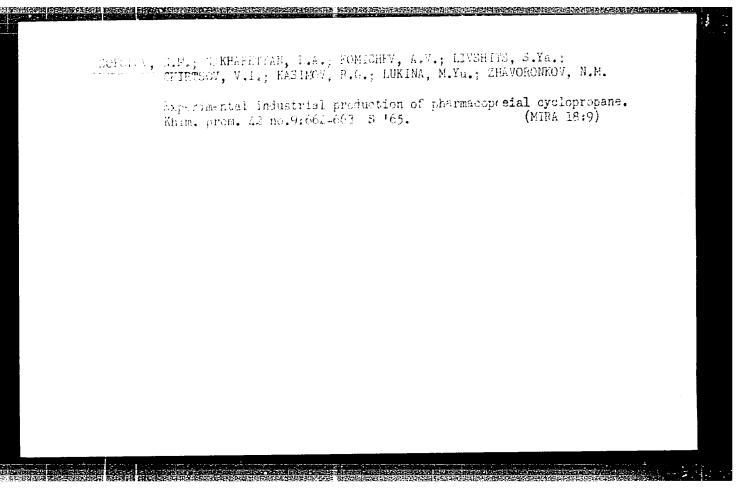
Third Working Conference on Stable Isotopes, 28 October to 2 November 1963, Leipzig.

UVAROV, O.V.; SOKOLOV, N.M.

Effect of the evaporation conditions on the value of the partition factor & in the course of Raleigh distillation. Zhur. fiz. khim. 38 no.7:1863-1864 Jl '64. (MIRA 18:3)

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AUTHOR: Sokolov, N. M.; Chirtsov, V. I.; Kasimov ORG: None	Nakhapetyan, L. A.; R. G.; Lukina, M.	CE CODE: UR/0064/65/000/009 Fomichev, A. V.; Livshits, Yu.; Zhavoronkov, N. M. of pharmacopoeial cycloprop	<u>s. ya.;</u> 54 1 B
SOURCE: Khimicheskaya p	promyshlennost, no.	9, 1965, 22-23	
propane	ne, organic synthetic #1,55 a1 cyclopropane was a CH2Cl + HBr -> BrCH	e process, cyclic group, phosynthesized via the following 2—CH2—CH2Cl	ermaceutical, ng steps: (1)
BrCH ₂ —	CH_2 — $CH_2C1 + Zn \rightarrow C$	H ₂ —CH ₂ + ZnClBr	(2)
packed tower. The oper process is described an cyclopropane is also il	d its diagram is giv	rities are removed by distilental industrial assembly usen. The reactor for the sympleme content of cyclopropare 661.715.4:547.512	nthesis or
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determined by gas-liquid the cyclopropane obtains permitted the refinement	d chromatography with a thermal conductivity de ed was found to meet the specified requirements t of certain parameters of the process by which the various stages, and improved the flowshee erably. Orig. art. has: 3 figures.	cyclo-
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SOKOLOV, N.M. (Gadyach)

Technic in intrabronchial administration of drugs. Vest.oto.-rin.
20 no.4:106 Jl-Ag '58 (MIRA 11:7)

(IMJECTIONS, BRONCHIAL)

SOKOLOV, N.M.

Diagnostic significance of an examination of bronchial irrigation waters used in pulmonary tuberculosis under sanatorium conditions.

Vrach. delo no.1:89 '59. (MIRA 12:4)

1. Gadyachskiy klinicheskiy sanatoriy. (TUBERCULOSIS--DIAGNOSIS)

